

Anoka-Hennepin Secondary Curriculum Unit Plan

Department:	Science	Course:	Chemistry I	Unit 5 Title:	Scientific Measurement	Grade Level(s):	10th
Assessed Trimester:	Trimester A	Pacing:	3 - 5 Days	Date Created:	6/7/2012	Last Revision Date:	6/23/2014

Course Understandings: *Students will understand that:*

- Problems can be solved and knowledge gained in a systematic way: solutions to one problem can create new questions and problems.
- Chemistry is recognized as significant in its application to other disciplines and the world.
- Ideas are expressed symbolically, numerically, and graphically.
- Behavior and properties of materials are organized, classified, and predicted utilizing periodic trends.
- Mathematical relationships are interpreted and manipulated to model the real world.
- The basic building blocks combine and recombine in a variety of ways to make all matter from the simple to the complex.
- The laws of chemistry predict outcomes that impact and apply to daily life.

DESIRED RESULTS (Stage 1) - WHAT WE WANT STUDENT TO KNOW AND BE ABLE TO DO?

Established Goals	
<ul style="list-style-type: none"><li>• <b>Standard:</b> <b>9C.1.3.4.1:</b> Use significant figures and an understanding of accuracy and precision in scientific measurements to determine and express the uncertainty of a result. <b>9C.1.3.4.1:</b> Use significant figures and an understanding of accuracy and precision in scientific measurements to determine and express the uncertainty of a result.</li><li>• <b>Literacy Standards:</b> <b>9.13.6.6:</b> Analyze the author’s purpose in describing phenomena, providing an explanation, describing a procedure, or discussing/reporting an experiment in a text, defining the question the author seeks to address. <b>9.13.9.9:</b> Compare and contrast findings presented in a text to those from other sources (including their own experiments), noting when the findings support or contradict previous explanations or accounts. (discuss models) <b>9.14.4.4:</b> Produce clear and coherent writing in which the development, organization, and style are appropriate to discipline, task, purpose, and audience. <b>9.14.5.5:</b> Use a writing process to develop and strengthen writing as needed by planning, drafting, revising, editing, rewriting, or trying a new approach, focusing on addressing what is most significant for a specific purpose and audience, and appropriate to the discipline. <b>9.14.6.6:</b> Use technology, including, but not limited to, the Internet, to produce, publish, and update individual or shared writing products and multi-media texts, taking advantage of technology’s capacity to link to other information and to display information flexibly and dynamically. <b>9.14.10.10:</b> Write routinely over extended time frames (time for reflection and revision) and shorter time frames (a single sitting or a day or two) for a range of discipline-specific tasks, purposes and audiences.</li></ul>	
Transfer	
<b>Students will be able to independently use their learning to: (product, high order reasoning)</b> <ul style="list-style-type: none"><li>• Use the concepts to determine if a number is appropriately reported.</li><li>• Explain how the concepts of accuracy and precision influence the answers to quantitative scientific questions.</li><li>• Why certain units are used to report and measure certain values.</li></ul>	
Meaning	
<b>Unit Understanding(s):</b> <b>Students will understand that:</b> <ul style="list-style-type: none"><li>• Measurements are not perfect and are subject to the accuracy and precision of the device(s) used.</li><li>• Calculations with measured values must be rounded to an appropriate and truthful value.</li></ul>	<b>Essential Question(s):</b> <b>Students will keep considering:</b> <ul style="list-style-type: none"><li>• Why are careful measurements and calculations essential to success in scientific fields?</li><li>• Why can’t I just write down the answer to two decimal places?</li></ul>

Acquisition	
<b>Knowledge - Students will:</b> <ul style="list-style-type: none"><li>Know rules for determining significant figures in measurements and calculations. 9C.1.3.4.1 (L)</li></ul>	<b>Reasoning - Students will:</b> <ul style="list-style-type: none"><li></li></ul> <b>Skills - Students will:</b> <ul style="list-style-type: none"><li>Students will use scientific notation. 9C.1.3.4.1 (L)</li><li>Express a measurement in terms of the uncertainty of the measurement. 9C.1.3.4.1 (L)</li></ul>

<b>Common Misunderstandings</b> <ul style="list-style-type: none"><li>Measurement is only linear.</li><li>Any quantity can be measured as accurately as you want.</li><li>Children who have used measuring devices at home already know how to measure.</li><li>The metric system is more accurate than other measurement systems (such as the English system).</li><li>The English system is easier to use than the metric system.</li><li>You can only measure to the smallest unit shown on the measuring device.</li><li>Some objects cannot be measured because of their size or inaccessibility.</li><li>The five senses are infallible.</li><li>An object must be “touched” to be measured.</li><li>A measuring device must be a physical object.</li><li>Mass and weight are the same and they are equal at all times.</li><li>Mass and volume are the same.</li><li>There is only one way to measure perimeter.</li><li>Only the area of rectangular shapes can be measured in square units.</li><li>You cannot measure the volume of some objects because they do not have “regular” lengths, widths, or heights.</li><li>An object’s volume is greater in water than in air.</li><li>The density of an object depends only on its volume.</li><li>Density for a given volume is always the same.</li></ul>	<b>Essential new vocabulary</b> <ul style="list-style-type: none"><li>Significant figures</li><li>Uncertainty</li><li>Accuracy</li><li>Precision</li></ul>
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